

## Blood Glucose Content Assay Kit

**Note:** The reagents of this product are subject to change. Please note and strictly follow this instruction.

**Operation Equipment:** Spectrophotometer

**Catalog Number:** BC2490

**Size:** 50T/48S

### Components:

**Reagent I:** 10 mL×1, 1 mmol/mL glucose solution. Storage at 2-8°C.

**Reagent II:** Liquid 25 mL×1. Storage at 2-8°C.

**Reagent III:** Liquid 25 mL×1. Storage at 2-8°C.

**Preparation of mixed reagent:** mix Solution II and Solution III in equal proportion and prepare it fresh.

### Product Description

Glucose in the blood of mammals is the main form of sugar transport in the body. Blood glucose concentration is regulated by the nervous system and hormones, so it remains relatively stable. While hyperglycemia and hypoglycemia occur when the regulation is out of balance. Hyperglycemia can be caused by diabetes, increased intracranial pressure and dehydration. After the meal, mental tension can also appear physiological high blood glucose. In contrast, hypoglycemia can occur in patients with such conditions as islet cell proliferation or cancer, hypophysis, adrenal cortex and hypothyroidism, and severe liver disease. In addition, hunger and strenuous exercise can cause temporary hypoglycemia.

Glucose oxidase can catalyze the oxidation of glucose to gluconic acid, and produce hydrogen peroxide. Peroxidase catalyzes the oxidation of 4-aminoantipyrine bisphenol by hydrogen peroxide to form colored compounds with characteristic absorption peaks at 505 nm.

### Reagents and Equipment Required but Not Provided.

Water-bath/constant temperature incubator, spectrophotometer, 1 mL glass cuvette, transferpettor and distilled water.

### Procedure

#### I. Sample preparation:

Mix 150  $\mu$ L of serum (plasma) with 150  $\mu$ L of distilled water, boil for 10 min in a boiling water bath (cover tightly to prevent water loss), cool to room temperature, then centrifuge at 8000g for 10 min at 25°C and reserve the supernatant (equivalent to the serum (plasma) being diluted 2 times).

Note: If the measurement result is small, the ratio of serum to distilled water can be adjusted (For example. 200  $\mu$ L of serum (slurry) mixed with 100  $\mu$ L of distilled water and boiled, that is diluted 1.5 times); if the measurement result is large, the supernatant can be diluted with distilled water.

#### II. Determination Procedure

1. Preheat the spectrophotometer for more than 30 min, adjust the wavelength to 505 nm, and

adjust to zero with distilled water.

2. Sample table (add Reagent in the EP tube):

Reagent (μL)	Blank Tube (B)	Standard Tube (S)	Test Tube (T)
Sample	-	-	100
Reagent I	-	100	-
distilled water	100	-	-
Mixed reagent	900	900	900

Mix thoroughly, 37°C water bath, keep warm for 15 min, read the absorbance of wavelength at 505 nm. The absorbance is named  $A_B$ ,  $A_S$  and  $A_T$ . The standard tube and blank tube only need to be measured 1-2 times.

### Calculation of blood glucose content:

$$\begin{aligned} \text{Blood glucose content } (\mu\text{mol/mL}) &= C_S \times (A_T - A_B) \div (A_S - A_B) \times F \\ &= 2 \times (A_T - A_B) \div (A_S - A_B) \end{aligned}$$

$C_S$ : concentration of standard solution, 1 μmol/mL;

$F$ : dilution multiple of serum (plasma) in pre-treatment, 2.

### Note:

If  $(A_T - A_B)$  is less than 0.01, the ratio of serum to distilled water can be adjusted (For example. 200 μL of serum (plasma) mixed with 100 μL of distilled water and boiled, that is. being diluted 1.5 times);  $(A_T - A_B)$  is greater than 1.5, the supernatant can be diluted with distilled water. Note the modification of the dilution times in the calculation formula.

### Experimental example:

1. 150 μL of sheep serum and 150 μL of distilled water were mixed and boiled for 10 min, the supernatant was centrifuged and determined according to the assay procedure, and the absorbance values were measured using a glass cuvette as  $A_T = 0.241$ ,  $A_B = 0.006$ ,  $A_S = 0.600$ . calculation

$$\begin{aligned} &\text{Blood glucose content} \\ (\mu\text{mol/mL}) &= C_S \times (A_T - A_B) \div (A_S - A_B) \times F = 2 \times (0.24 - 0.006) \div (0.600 - 0.006) = 0.791 \mu\text{mol/mL} \end{aligned}$$

2. 150 μL of cow serum and 150 μL of distilled water were mixed and boiled for 10 min, the supernatant was centrifuged and then diluted 2 times with distilled water (the overall dilution was 4 times) and then measured according to the assay procedure, the absorbance value of  $A_T = 1.047$ ,  $A_B = 0.006$  and  $A_S = 0.600$  measured with a glass cuvette. calculation

$$\text{Blood glucose content } (\mu\text{mol/mL}) = 1 \times (1.047 - 0.006) \div (0.600 - 0.006) \times 4 = 7.010 \mu\text{mol/mL}.$$

### Recent Product Citations:

[1] Wu J, Liu J, Ding Y, et al. MiR-455-3p suppresses renal fibrosis through repression of ROCK2 expression in diabetic nephropathy[J]. Biochemical and biophysical research communications, 2018, 503(2): 977-983.

### References:

[1]Basagni U, Bonicolini F. Ready to use liquid reagent for determining the glucose content in blood: U.S. Patent 5,077,199[P]. 1991-12-31.

[2] Kabasakalian P, Kalliney S, Westcott A. Enzymatic blood glucose determination by colorimetry of N, N-diethylaniline-4-aminoantipyrine[J]. Clinical chemistry, 1974, 20(5): 606-607.

**Related Products:**

BC0340/BC0345	Glucogen Content Assay Kit
BC2540/BC2545	Cellulase(CL) Activity Assay Kit
BC0330/BC0335	Trehalose Content Assay Kit
BC2500/BC2505	Glucose Content Assay Kit

**Technical Specifications:**

The detection limit: 0.0078  $\mu\text{mol/mL}$

Linear range: 0.0625-3  $\mu\text{mol/mL}$