

Hsphasetin (HP) Activity Assay Kit

Note: It is necessary to predict 2-3 large difference samples before the formal determination.

Operation Equipment: Spectrophotometer

Catalog Number: BC4340

Size: 50T/24S

Components:

Reagent I: Liquid 20mL×1, store at 4°C.

Reagent II: Liquid 3mL×1, store at 4°C.

Reagent III: Powder×1, store at 4°C. Add 7 mL of distilled water to dissolve the reagent before use. Unused reagent is still stored at 4°C for four weeks.

Standard: Liquid 1mL ×1, store at 4°C. 9 μmol/mL Fe²⁺ standard solution.

Product Description:

Hsphasetin (HP) is a homologue of ceruloplasmin, which catalyzes the oxidation of ferrous ions (Fe²⁺) to ferric ions (Fe³⁺), then Fe³⁺ binds to transferrin and participates in cellular iron release.

With Fe²⁺ as substrate, Fe²⁺ is oxidized to Fe³⁺ under the catalysis of HP. Fe²⁺ forms a colored complex with phenazine, and has a characteristic absorption peak at 562 nm. The content of Fe²⁺ which is not oxidized is calculated, and then the content of oxidized Fe²⁺ is obtained. So the HP activity can be reflected by the rate at Fe²⁺ oxidized.

Reagents and Equipment Required but Not Provided:

Spectrophotometer, adjustable transferpettor, balance, mortar/homogenizer, centrifuge, 1mL glass cuvette and distilled water.

Sample preparation:

1. Plant and animal tissues: Plant and animal tissues: mass (g): the volume of distilled water (mL) is 1: 5 ~ 10, weigh about 0.1 g of sample, add 1 mL of distilled water, Ice bath homogenate and fully grind. Centrifuge 10000 rpm at 4°C for 10 min, Take the supernatant on ice for testing.
2. Serum or culture medium: It is recommended to dilute serum or plasma 2-4 times with distilled water and directly test.

Determination procedure:

1. Preheat spectrophotometer for 30min, adjust the wavelength to 562 nm and set counter to zero with distilled water.
2. Standard working solution: dilute 9μmol/mL NaNO₂ standard solution with distilled water to 360, 180, 90, 45, 22.5, 11.25, 5.625 nmol/mL for use.
3. Add reagent as follows:

| Reagent (μL) | Control tube | Test tube | Matrix-free tube | Blank tube | Standard |
|--------------|--------------|-----------|------------------|------------|----------|
|--------------|--------------|-----------|------------------|------------|----------|

| | (Ac) | (At) | (Am) | (Ab) | tube (As) |
|---|------|------|------|------|-----------|
| Distilled water | - | - | 20 | 20 | 20 |
| Sample | 20 | 20 | - | - | - |
| Reagent I | 280 | 280 | 280 | 280 | 280 |
| Use a pipette to blow and mix thoroughly | | | | | |
| Reagent II | - | 100 | 100 | - | - |
| Standard | - | - | - | - | 100 |
| Distilled water | 100 | - | - | 100 | - |
| Mix well, accurately react in a 37 ° C water bath or constant temperature incubator for 3 min | | | | | |
| Reagent III | 100 | 100 | 100 | 100 | 100 |
| Distilled water | 500 | 500 | 500 | 500 | 500 |

Mix and measure the absorbance at 562 nm in the 1 mL glass cuvette, record it as Ac, At, Am, Ab and As. Calculate $\Delta A = (Am - Ab) - (At - Ac)$, $\Delta A_s = As - Ab$. Each test tube should be provided with one contrast tube. Standard curve and blank tube only need to be measured once or twice.

Calculation:

1. Standard curve drawing

Taking the concentration of each standard solution as the x-axis and its corresponding ΔA_s as the y-axis, draw a standard curve to get the standard equation $y = kx + b$, and bring ΔA_T into the equation to get x ($\mu\text{mol/mL}$).

2. Protein concentration:

Unit definition: One unit of enzyme activity is defined as the amount of enzyme that catalyzes the oxidation of 1 nmol of Fe^{2+} per minutes every milligram tissue protein in the reaction system.

$$\text{HP (U/mg prot)} = x \times V_2 \div (V_s \times C_{pr}) \div T \times N = 1.667x \div C_{pr} \times N$$

3. Sample weight:

Unit definition: One unit of enzyme activity is defined as the amount of enzyme that catalyzes the oxidation of 1 nmol of Fe^{2+} per min every gram tissue in the reaction system.

$$\text{HP (U/g)} = x \times V_2 \div (V_s \times W \div V_e) \div T \times N = 1.667x \div W \times N$$

4. Liquid volume:

Unit definition: One unit of enzyme activity is defined as the amount of enzyme that catalyzes the oxidation of 1 nmol of Fe^{2+} every milliliter liquid in the reaction system per min.

$$\text{HP (U/mL)} = x \times V_2 \div V_s \div T \times N = 1.667x \times N$$

V_2 : the volume of reagent II added, 0.1mL.

V_s : sample volume added, 0.002mL;

V_e : volume used in the extraction solution, 1mL;

C_{pr} : sample protein concentration, mg/mL;

W : Fresh weight of sample, g;

T: React time, 3min.

N: Dilution factor.

Note:

1. When the determination of A is less than 0.1, it is recommended to dilute the crude enzyme solution with distilled water before performing the measurement, and multiply the dilution factor in the calculation formula.

Experimental Examples:

1. Take 0.1g of spleen and add 1mL extract to homogenize and grind, take the supernatant and operate according to the measurement steps, and calculate $\Delta A = (A_m - A_b) - (A_t - A_c) = (0.976 - 0.005) - (0.996 - 0.047) = 0.022$, bring standard curve line $y = 0.0027x + 0.0036$, $x = 6.815$, calculate the enzyme based on the sample weight:

$$HP (U/g) = 1.667 \times x \div W \times N = 1.667 \times 6.815 \div 0.1 = 113.603 \text{ U/g weight}$$

2. Take rabbit serum and operate according to the measurement steps, calculate $\Delta A = (A_m - A_b) - (A_t - A_c) = (0.976 - 0.005) - (0.469 - 0.009) = 0.511$, bring standard curve line $y = 0.0027x + 0.0036$, $x = 187.926$, calculate the enzyme based on the liquid volume:

$$HP (U/mL) = 1.667 \times x = 1.667 \times 187.926 = 313.273 \text{ U/mL}$$

Related Products:

BC1300/BC1305 Ceruloplasmin(CP) Assay Kit

BC1310/BC1315 Total antioxidant capacity(T-AOC) Assay Kit

BC4430/BC4435 Uricase Activity Assay Kit

BC1360/BC1365 Uric acid (UA) Assay Kit

BC1320/BC1325 Hydroxyl Free Radical Scavenging Capacity Assay Kit