

## Hsphasetin (HP) Activity Assay Kit

**Note:** It is necessary to predict 2-3 large difference samples before the formal determination.

**Operation Equipment:** Spectrophotometer/ Microplate Reader

**Catalog Number:** BC4345

**Size:** 100T/48S

### Components:

**Reagent I:** Liquid 20mL×1, store at 4°C.

**Reagent II:** Liquid 3mL×1, store at 4°C.

**Reagent III:** Powder×1, store at 4°C. Add 6 mL of distilled water to dissolve the reagent before use. Unused reagent is still stored at 4°C for four weeks.

**Standard:** Liquid 1mL ×1, store at 4°C. 9 μmol/mL Fe<sup>2+</sup> standard solution.

### Product Description:

Hsphasetin (HP) is a homologue of ceruloplasmin, which catalyzes the oxidation of ferrous ions (Fe<sup>2+</sup>) to ferric ions (Fe<sup>3+</sup>), then Fe<sup>3+</sup> binds to transferrin and participates in cellular iron release.

With Fe<sup>2+</sup> as substrate, Fe<sup>2+</sup> is oxidized to Fe<sup>3+</sup> under the catalysis of HP. Fe<sup>2+</sup> forms a colored complex with phenazine, and has a characteristic absorption peak at 562 nm. The content of Fe<sup>2+</sup> which is not oxidized is calculated, and then the content of oxidized Fe<sup>2+</sup> is obtained. So the HP activity can be reflected by the rate at Fe<sup>2+</sup> oxidized.

### Reagents and Equipment Required but Not Provided:

Spectrophotometer/microplate reader, adjustable transferpettor, balance, mortar/homogenizer, centrifuge, micro glass cuvette/ 96 well flat-bottom plate and distilled water.

### Sample preparation:

1. Plant and animal tissues: Plant and animal tissues: mass (g): the volume of distilled water (mL) is 1: 5 ~ 10, weigh about 0.1 g of sample, add 1 mL of distilled water, Ice bath homogenate and fully grind. Centrifuge 10000 rpm at 4°C for 10 min, Take the supernatant on ice for testing.
2. Serum or culture medium: It is recommended to dilute serum or plasma 2-4 times with distilled water and directly test.

### Determination procedure:

1. Preheat spectrophotometer/ microplate reader for 30min, adjust the wavelength to 562 nm and set spectrophotometer counter to zero with distilled water.
2. Standard working solution: dilute 9μmol/mL NaNO<sub>2</sub> standard solution with distilled water to 360, 180, 90, 45, 22.5, 11.25, 5.625 nmol/mL for use.
3. Add reagent as follows:

Reagent (μL)	Control tube	Test tube	Matrix-free tube	Blank tube	Standard
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	(Ac)	(At)	(Am)	(Ab)	tube (As)
Distilled water	-	-	8	8	8
Sample	8	8	-	-	-
Reagent I	112	112	112	112	112
Use a pipette to blow and mix thoroughly					
Reagent II	-	40	40	-	-
Standard	-	-	-	-	40
Distilled water	40	-	-	40	-
Mix well, accurately react in a 37 ° C water bath or constant temperature incubator for 3 min					
Reagent III	40	40	40	40	40

Mix and measure the absorbance at 562 nm in the micro glass cuvette/ 96 well plate, record it as Ac, At, Am, Ab and As. Calculate  $\Delta A = (Am - Ab) - (At - Ac)$ ,  $\Delta A_s = As - Ab$ . Each test tube should be provided with one contrast tube. Standard curve and blank tube only need to be measured once or twice.

### Calculation:

#### 1. Standard curve drawing

Taking the concentration of each standard solution as the x-axis and its corresponding  $\Delta A_s$  as the y-axis, draw a standard curve to get the standard equation  $y = kx + b$ , and bring  $\Delta A_T$  into the equation to get x ( $\mu\text{mol/mL}$ ).

#### 2. Protein concentration:

Unit definition: One unit of enzyme activity is defined as the amount of enzyme that catalyzes the oxidation of 1 nmol of  $\text{Fe}^{2+}$  per minutes every milligram tissue protein in the reaction system.

$$\text{HP (U/mg prot)} = x \times V_2 \div (V_s \times C_{pr}) \div T \times N = 1.667x \div C_{pr} \times N$$

#### 3. Sample weight:

Unit definition: One unit of enzyme activity is defined as the amount of enzyme that catalyzes the oxidation of 1 nmol of  $\text{Fe}^{2+}$  per min every gram tissue in the reaction system.

$$\text{HP (U/g)} = x \times V_2 \div (V_s \times W \div V_e) \div T \times N = 1.667x \div W \times N$$

#### 4. Liquid volume:

Unit definition: One unit of enzyme activity is defined as the amount of enzyme that catalyzes the oxidation of 1 nmol of  $\text{Fe}^{2+}$  every milliliter liquid in the reaction system per min.

$$\text{HP (U/mL)} = x \times V_2 \div V_s \div T \times N = 1.667x \times N$$

$V_2$ : the volume of reagent II added, 0.04mL.

$V_s$ : sample volume added, 0.008mL;

$V_e$ : volume used in the extraction solution, 1mL;

$C_{pr}$ : sample protein concentration, mg/mL;

$W$ : Fresh weight of sample, g;

$T$ : React time, 3min.

N: Dilution factor.

**Note:**

1. When the determination of A is less than 0.1, it is recommended to dilute the crude enzyme solution with distilled water before performing the measurement, and multiply the dilution factor in the calculation formula.

**Experimental Examples:**

1. Take 0.1g of colon and add 1mL extract to homogenize and grind, take the supernatant and operate according to the measurement steps, and calculate  $\Delta A = (A_m - A_b) - (A_t - A_c) = (1.144 - 0.041) - (1.114 - 0.048) = 0.037$ , bring standard curve line  $y = 0.0031x - 0.0007$ ,  $x = 11.710$ , calculate the enzyme based on the sample weight:

$$HP (U/g) = 1.667 \times x \div W \times N = 1.667 \times 11.710 \div 0.1 = 195.161 \text{ U/g weight}$$

2. Take rabbit serum and operate according to the measurement steps, calculate  $\Delta A = (A_m - A_b) - (A_t - A_c) = (0.577 - 0.048) - (1.114 - 0.048) = 0.574$ , bring standard curve line  $y = 0.0031x - 0.0007$ ,  $x = 184.935$ , calculate the enzyme based on the liquid volume:

$$HP (U/mL) = 1.667 \times x = 1.667 \times 184.935 = 308.287 \text{ U/mL}$$

**Related Products:**

BC1300/BC1305 Ceruloplasmin(CP) Assay Kit

BC1310/BC1315 Total antioxidant capacity(T-AOC) Assay Kit

BC4430/BC4435 Uricase Activity Assay Kit

BC1360/BC1365 Uric acid (UA) Assay Kit

BC1320/BC1325 Hydroxyl Free Radical Scavenging Capacity Assay Kit